

PREPARATION OF 1-ARYL-2-(BENZYLTHIO)-(1,2-DIDEOXY-D-*glycero*- β -L-*gluco*-HEPTOFURANO)[1,2-*d*]-2-IMIDAZOLINES, AND NEW, ACYCLIC C-NUCLEOSIDES OF THE IMIDAZOLE*

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(Received July 18th, 1983; accepted for publication, September 20th, 1983)

ABSTRACT

The reaction of 1-aryl-(1,2-dideoxy-D-*glycero*- β -L-*gluco*-heptofurano)[1,2-*d*]imidazolidine-2-thiones with benzyl chloride and an equivalent amount of sodium hydrogencarbonate yields 1-aryl-2-(benzylthio)-(1,2-dideoxy-D-*glycero*- β -L-*gluco*-heptofurano)[1,2-*d*]-2-imidazolines (**2**). If the reaction is carried out in the absence of sodium hydrogencarbonate, the 1-aryl-2-(benzylthio)-4-(D-*galacto*-pentitol-1-yl)imidazoles are obtained. These compounds are also obtained by acid-catalyzed isomerization of compounds **2**.

INTRODUCTION

The reaction of amino sugars with alkyl and aryl isothiocyanates yields 1-alkyl(aryl)-(1,2-dideoxy-D-*glyco*furano)[1,2-*d*]imidazolidine-2-thiones¹ which are useful intermediates in the synthesis of imidazole derivatives having interesting pharmacological and biological properties^{2,3}. We have recently described⁴ the preparation of 1-aryl-(1,2-dideoxy-D-*glycero*- β -L-*gluco*-heptofurano)[1,2-*d*]imidazolidine-2-thiones (**1**) by reaction between aryl isothiocyanates and 2-amino-2-deoxy-D-*glycero*-L-*gluco*-heptose. We now report the preparation of some of their 2-benzylthio derivatives, and the isomerization to acyclic C-nucleosides of the imidazole.

RESULTS AND DISCUSSION

The reaction of 1-aryl-(1,2-dideoxy-D-*glycero*- β -L-*gluco*-heptofurano)[1,2-*d*]imidazolidine-2-thiones (**1**) with benzyl chloride and the equivalent amount of sodium hydrogencarbonate yields 1-aryl-2-(benzylthio)-(1,2-dideoxy-D-*glycero*- β -L-*gluco*-heptofurano)[1,2-*d*]-2-imidazolines (**2**). The structures of these compounds

*Taken, in part, from the Ph. D. Thesis of F.R.V.

TABLE I
¹H-N.M.R. DATA ^a (90 MHz) FOR **3a**, **3b**, AND **3c**

Com- pound No.	H-1'	H-2'	H-3' ^b	H-4' ^b	H-5' ^b	H-6'	H-7' ^b	H-7' ^{bb}	OAc	S-CH ₂	phenyl	p-tolyl	p-bromophenyl
3a	5.78 d	4.57 d	5.52 d	3.87 dd	5.34 dd	5.35-5.60 m	4.35 dd	3.95 dd	1.94 s (3 H)	4.32 s (2 H)	7.60-7.15 m (10 H)		
	<i>J</i> _{1',2'} 5.9	<i>J</i> _{2',3'} 0.0	<i>J</i> _{3',4'} 2.6	<i>J</i> _{4',5'} 9.6	<i>J</i> _{5',6'} 2.6	<i>J</i> _{6',7'} 5.0	<i>J</i> _{7',7''} 11.6		2.02 s (3 H)				
3b	5.72 d	4.54 d	5.53 d	3.92 dd	5.47 dd	5.35-5.60 m	4.37 dd	3.96 dd	2.09 s (6 H)	4.32 s (2 H)	7.50-7.20 m (5 H)	2.35 s (3 H)	
	<i>J</i> _{1',2'} 5.6	<i>J</i> _{2',3'} 0.0	<i>J</i> _{3',4'} 2.3	<i>J</i> _{4',5'} 9.6	<i>J</i> _{5',6'} 2.6	<i>J</i> _{6',7'} 5.0	<i>J</i> _{7',7''} 11.2		2.04 s (3 H)			7.15 s (4 H)	
3c	5.72 d	4.56 d	5.52 d	3.85 dd	5.30 dd	5.33-5.60 m	4.36 dd	3.94 dd	2.10 s (6 H)	4.30 s (2 H)	7.40-7.15 m (5 H)		7.55-7.05 m (4 H)
	<i>J</i> _{1',2'} 5.6	<i>J</i> _{2',3'} 0.0	<i>J</i> _{3',4'} 2.8	<i>J</i> _{4',5'} 9.5	<i>J</i> _{5',6'} 2.6	<i>J</i> _{6',7'} 5.0	<i>J</i> _{7',7''} 11.4		2.02 s (3 H)				
						<i>J</i> _{6',7''} 7.3			2.09 s (6 H)				

^aRecorded in CDCl₃ at 35.5°, δ scale (internal Me₄Si), *J* in Hz. ^bValues obtained by extrapolation back to zero concentration of Eu(fod)₃.

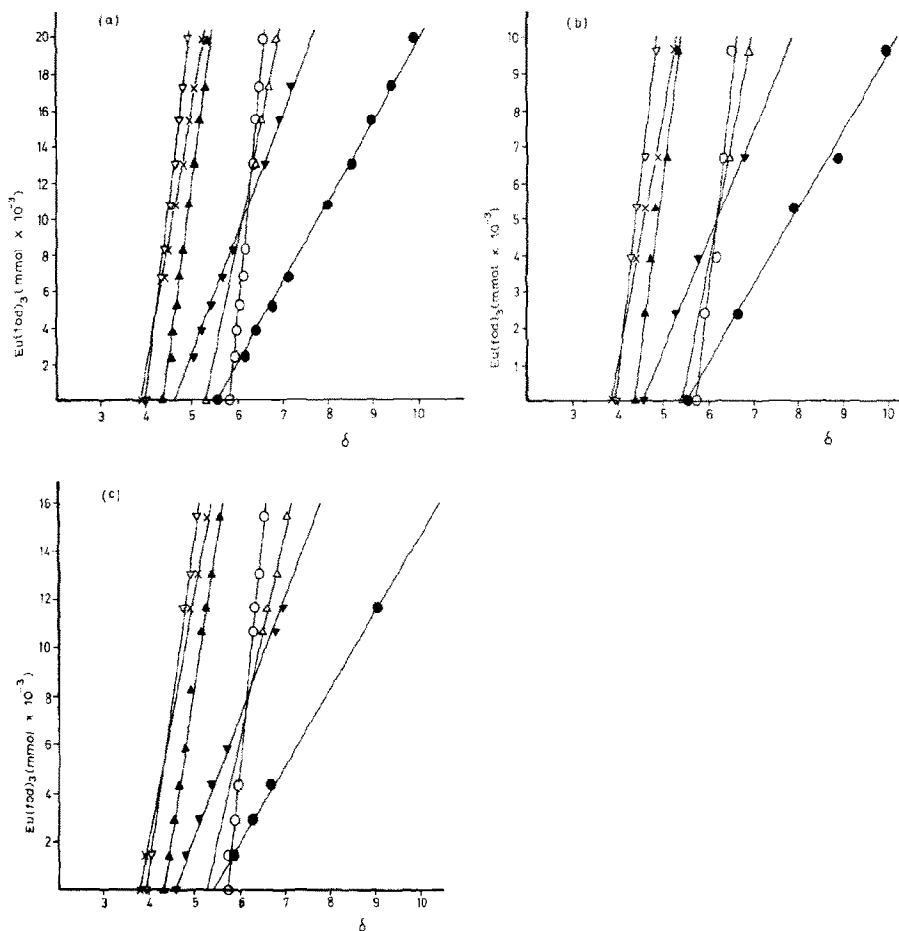


Fig. 1. Plot of the chemical shifts of the proton resonances of (a) 2-(benzylthio)-1-phenyl-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero-β-L-gluco-heptofurano)[1,2-*d*]-2-imidazoline (3a); (b) 2-(benzylthio)-1-*p*-tolyl-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero-β-L-gluco-heptofurano)[1,2-*d*]-2-imidazoline (3b); and (c) 2-(benzylthio)-1-(*p*-bromophenyl)-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero-β-L-gluco-heptofurano)[1,2-*d*]-2-imidazoline (3c), vs. the mmol of Eu(fod)₃ added. Values for the unperturbed chemical shifts were estimated by extrapolation back to zero concentration of Eu(fod)₃. O, H-1'; ▼, H-2'; ●, H-3'; ×, H-4'; Δ, H-5'; ▲, H-7'; ▽, H-7'.

were demonstrated by elemental analyses and spectral data (u.v. and i.r.). The structure of the glycofuranoid ring was proved by periodate oxidation, -1 molar equivalent of formic acid being produced in each case.

Conventional treatment of **2** with acetic anhydride in pyridine gave the corresponding tetra-*O*-acetyl derivatives (**3**), whose ¹H-n.m.r.-spectral data are collected in Table I. Complete sets of proton-proton spin-coupling constants were determined for compounds **3** in the presence of Eu(fod)₃; chemical values were esti-

TABLE II

VALUES OF J (Hz) AND ϕ (DEGREES) FOR THE FURANOID-RING PROTONS

i, j	J_{ij}	$\phi_{i,j}$ (calculated) ⁷	$\phi_{i,j}$ (taken from models) ^{8,9}		
			4T_3	E_3	4E
1,2	5.9	31	19	37	0
2,3	0.0	78	70	60	83
3,4	2.6	53	60	60	60

mated by extrapolation of the approximately linearly dependent, paramagnetically altered, field positions to zero concentration of added lanthanide, as illustrated in Fig. 1. The J values in the absence of $\text{Eu}(\text{fod})_3$ may be taken to be almost the same as these values, as the coupling constants are generally not greatly influenced by addition of a shift reagent^{5,6}; in addition, the $J_{1',2'}$ and $J_{2',3'}$ values were found invariable before and after the addition of the shift reagent. The dihedral angles (ϕ) between all of the vicinal protons were calculated from these values of J by applying a modified Karplus equation⁷. By comparing these ϕ values with those obtained from models^{8,9} (see Table II), it may be assumed that the most favored conformation of the furanoid rings is intermediate between the E_3 and 4T_3 forms.

If the benzylation of **1** is conducted in the absence of sodium hydrogen-carbonate, isomerization of the glycosyl moiety occurs, and the 1-aryl-2-(benzyl-

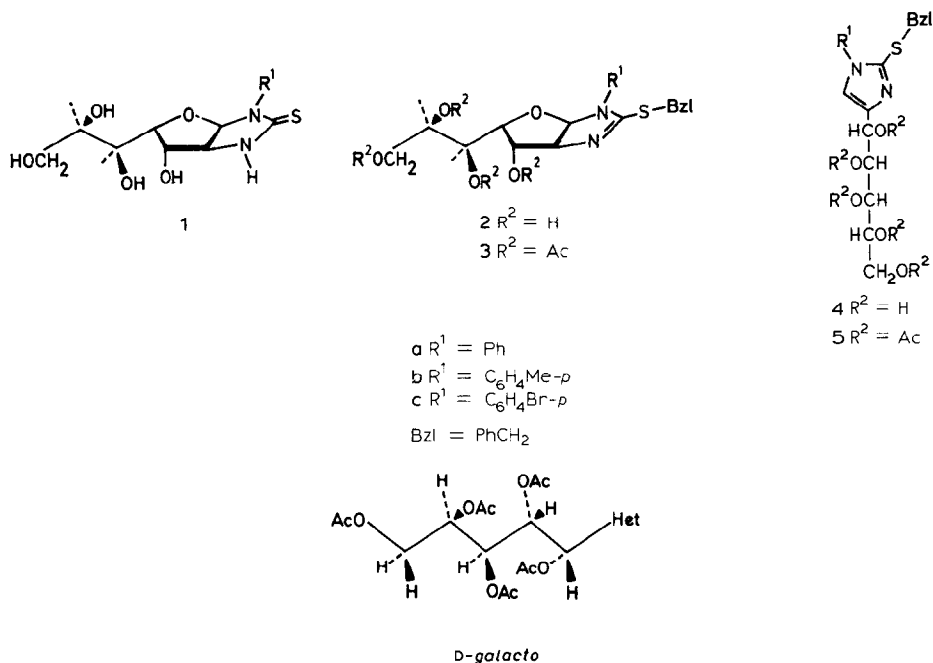


TABLE III
¹H-N.M.R. DATA^a (90 MHz) FOR 5a, 5b, AND 5c

Com- pound No.	H-1'	H-2'	H-3'	H-4'	H-5'	H-5''	OAc	H-5	S-CH ₂	phenyl	p-tolyl	p-bromophenyl
5a	6.09 d	5.76 dd	5.55 dd	5.37 m	4.33 dd	3.98 dd	2.05 s (3H)	7.04 s	4.26 s (2H)	6.95-7.50 m		
	J _{1',2'} 3.4	J _{2',3'} 8.7	J _{3',4'} 2.2	J _{4',5'} 4.9 J _{4',5''} 7.0	J _{5',5''} 11.6		2.09 s (3H)			7.22 s		
							2.12 s (6H)					
5b	6.06 d	5.71 dd	5.50 dd	5.32 m	4.30 dd	3.93 dd	2.14 s (3H)	6.98 s	4.23 s	7.20 s	6.80-7.30 m	
	J _{1',2'} 3.3	J _{2',3'} 8.3	J _{3',4'} 2.0	J _{4',5'} 4.6 J _{4',5''} 7.1	J _{5',5''} 11.4		2.01 s (3H)				2.35 s (3H)	
							2.05 s (3H)					
5c	6.03 d	5.70 dd	5.49 dd	5.33 m	4.31 dd	3.94 dd	2.08 s (6H)	6.97 s	4.21 s	7.18 s		6.75-7.55 m
	J _{1',2'} 3.1	J _{2',3'} 8.4	J _{3',4'} 2.1	J _{4',5'} 4.6 J _{4',5''} 7.3	J _{5',5''} 11.5		2.11 s (3H)					
							2.05 s (3H)					
							2.07 s (3H)					
							2.08 s (3H)					
							2.11 s (3H)					

^aRecorded in CDCl₃ at 35.5°, δ scale (internal Me₄Si), J in Hz.

thio)-4-(*D*-galacto-pentitol-1-yl)imidazoles (**4**) are obtained. These compounds are also obtained by trifluoroacetic acid-catalyzed isomerization of compounds **2**. The structures of compounds **4** were demonstrated by elemental analyses and spectral data (u.v. and i.r.). The presence of the pentahydroxypentyl chain was proved by periodate oxidation, showing a periodate consumption of 4 mol per mol of substance. Assignment of the *D*-galacto configuration is based on the configuration of the sugar precursor, and is consistent with the Richtmyer-Hudson rules¹⁰. The structure was also proved by preparation of their pentaacetates (**5**). The vicinal spin-coupling constants observed between H-1', H-2' and H-3', H-4' are of small magnitude (see Table III), indicating *gauche*-disposed protons, and the value of $J_{2',3'}$ indicated *trans*-periplanar protons. These values established that the molecules exist essentially in the fully extended, planar, zigzag conformation. The intermediate values of $J_{4',5'}$ and $J_{4',5''}$ (4.7 and 7.2 Hz) are in agreement with the chain-end flexibility encountered in other examples of acyclic-sugar derivatives^{11,12}.

EXPERIMENTAL

General methods. — Solutions were concentrated *in vacuo* at temperatures below 40°. Melting points were determined with a Gallenkamp apparatus, and are uncorrected. Optical rotations were measured at $20 \pm 2^\circ$ with a Perkin-Elmer 141 polarimeter (10-cm cell). I.r. spectra (KBr discs) were recorded with Beckman IR-33 and Perkin-Elmer 399 spectrophotometers, and u.v. spectra with a Beckman 25 instrument. ¹H-N.m.r. spectra (90 MHz, internal Me₄Si) were recorded with a Perkin-Elmer R-32 spectrometer, and coupling constants were measured directly from spectra recorded at 300-Hz sweep-width (temperature of the probe, 35.5°). Assignments were confirmed by double resonance, and overlapping signals were gradually shifted and separated from one another by incremental addition of Eu(fod)₃. T.l.c. was performed on silica gel GF₂₅₄ (Merck) with 3:1 ethyl acetate-ethanol and detection with u.v. light and iodine vapor. Consumption of periodate and formic acid produced were determined as previously described¹³⁻¹⁵.

2-(Benzylthio)-1-phenyl-(1,2-dideoxy-D-glycero-β-L-gluco-heptofurano)-[1,2-d]-2-imidazoline (2a). — A solution of 1-phenyl-(1,2-dideoxy-D-glycero-β-L-gluco-heptofurano)[1,2-d]imidazolidine-2-thione⁴ (**1a**; 1 g, 3 mmol) in 90% ethanol (10 mL) was treated with NaHCO₃ (0.26 g, 3 mmol) and benzyl chloride (0.4 mL, 3 mmol). The mixture was boiled under reflux for 1 h, cooled, and concentrated to two-thirds of its original volume, and the product (**2a**) then crystallized (1.042 g, 81%). Recrystallized from 1:1 ethanol-water, it had m.p. 176-177°, $[\alpha]_D^{25} -80.5^\circ$ (*c* 0.7, pyridine); $\lambda_{\max}^{96\% \text{ EtOH}}$ 250 nm (ϵ_{mM} 11.00); ν_{\max} 3500-3100 (OH), 2950, 2930 and 2865 (C-H), 1595, 1580, 1480 and 1440 cm⁻¹ (C=C aromatic).

Anal. Calc. for C₂₁H₂₄N₂O₅S: C, 60.56; H, 5.80; N, 6.72; S, 7.69. Found: C, 60.65; H, 6.01; N, 6.62; S, 7.67. Formic acid produced: 0.94 mol.

2-(Benzylthio)-1-phenyl-(3,5,6,7-tetra-O-acetyl-1,2-dideoxy-D-glycero-β-L-

gluco-heptofurano)[1,2-d]-2-imidazoline (**3a**). — Conventional treatment of **2a** (0.3 g, 0.72 mmol) with pyridine (3 mL) and acetic anhydride (2 mL) gave **3a** (0.366 g, 86%). Crystallized from ethanol it had m.p. 124–126°, $[\alpha]_D^{18}$ -82.3° (*c* 0.52, pyridine); $\lambda_{\max}^{96\% \text{ EtOH}}$ 248 nm (ϵ_{mM} 11.50); ν_{\max} 1730 (C=O), 1490 and 1445 (C=C aromatic), 1250 and 1220 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table I.

Anal. Calc. for $\text{C}_{29}\text{H}_{32}\text{N}_2\text{O}_9\text{S}$: C, 59.57; H, 5.51; N, 4.79; S, 5.48. Found: C, 59.60; H, 5.64; N, 4.77; S, 5.82.

2-(Benzylthio)-1-*p*-tolyl-(1,2-dideoxy-D-glycero- β -L-gluco-heptofurano)-[1,2-d]-2-imidazoline (**2b**). — A solution of 1-*p*-tolyl-(1,2-dideoxy-D-glycero- β -L-gluco-heptofurano)[1,2-d]imidazolidine-2-thione⁴ (**1b**; 1.02 g, 3 mmol) in 90% ethanol (10 mL) was treated with NaHCO_3 (0.26 g, 3 mmol) and benzyl chloride (0.4 mL, 3 mmol). The mixture was boiled under reflux for 1 h; **2b** crystallized when it was cooled to 0° (1.113 g, 88%). Recrystallized from ethanol, it had m.p. 188–189°, $[\alpha]_D^{22}$ -66.5° , $[\alpha]_{578}^{22}$ -69.6° , $[\alpha]_{546}^{22}$ -81.3° , $[\alpha]_{436}^{22}$ -126.4° , $[\alpha]_{365}^{22}$ -325.7° (*c* 0.51, pyridine); $\lambda_{\max}^{96\% \text{ EtOH}}$ 248 nm (ϵ_{mM} 12.50); ν_{\max} 3500–3100 (OH), 1595, 1500 and 1490 cm^{-1} (C=C aromatic).

Anal. Calc. for $\text{C}_{22}\text{H}_{26}\text{N}_2\text{O}_5\text{S}$: C, 61.40; H, 6.05; N, 6.51; S, 7.44. Found: C, 61.66; H, 6.30; N, 6.44; S, 7.42. Formic acid produced: 0.91 mol.

2-(Benzylthio)-1-*p*-tolyl-(3,5,6,7-tetra-O-acetyl-1,2-dideoxy-D-glycero- β -L-gluco-heptofurano)[1,2-d]-2-imidazoline (**3b**). — Conventional treatment of **2b** (0.129 g, 0.3 mmol) with pyridine (1.2 mL) and acetic anhydride (0.65 mL, 6.9 mmol) gave **3b** (0.157 g, 87%). Recrystallized from 96% ethanol, it had m.p. 147–148°, $[\alpha]_D^{22}$ -83.4° , $[\alpha]_{578}^{22}$ -87.2° , $[\alpha]_{546}^{22}$ -101.6° , $[\alpha]_{436}^{22}$ -194.2° , $[\alpha]_{365}^{22}$ -364.8° (*c* 0.54, pyridine); $\lambda_{\max}^{96\% \text{ EtOH}}$ 246 nm (ϵ_{mM} 11.50); ν_{\max} 1735 (C=O), 1460 and 1440 (C=C aromatic), 1250 and 1220 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table I.

Anal. Calc. for $\text{C}_{30}\text{H}_{34}\text{N}_2\text{O}_9\text{S}$: C, 60.20; H, 5.68; N, 4.68; S, 5.35. Found: C, 59.93; H, 5.26; N, 4.61; S, 5.81.

2-(Benzylthio)-1-*p*-bromophenyl-(1,2-dideoxy-D-glycero- β -L-gluco-heptofurano)[1,2-d]-2-imidazoline (**2c**). — A solution of 1-(*p*-bromophenyl)-(1,2-dideoxy-D-glycero- β -L-gluco-heptofurano)[1,2-d]imidazolidine-2-thione⁴ (**1c**; 0.98 g, 2.4 mmol) in 90% ethanol (12 mL) was treated with NaHCO_3 (0.21 g, 2.4 mmol) and benzyl chloride (0.3 mL, 2.4 mmol). The mixture was boiled under reflux for 1.5 h, cooled to room temperature, and a precipitate of ionic salts filtered off; **2c** crystallized when the filtrate was cooled to 0° and water (1 mL) added; yield 0.751 g (68%). Recrystallized from ethanol, it had m.p. 199–201°, $[\alpha]_D^{26}$ -96.7° , $[\alpha]_{578}^{26}$ -101.8° , $[\alpha]_{546}^{26}$ -282.7° , $[\alpha]_{436}^{26}$ -247.6° , $[\alpha]_{365}^{26}$ -520.9° (*c* 0.55, pyridine); $\lambda_{\max}^{96\% \text{ EtOH}}$ 259 nm (ϵ_{mM} 12.60); ν_{\max} 3500–3000 (OH), 2960, 2940 and 2870 (C–H), 1595, 1580, 1485 and 1445 cm^{-1} (C=C aromatic).

Anal. Calc. for $\text{C}_{21}\text{H}_{23}\text{BrN}_2\text{O}_5\text{S}$: C, 50.91; H, 4.67; Br, 16.13; N, 5.65; S, 6.47. Found: C, 50.94; H, 4.77; Br, 16.11; N, 5.67; S, 6.67. Formic acid produced: 0.95 mol.

2-(Benzylthio)-1-*p*-bromophenyl-(3,5,6,7-tetra-O-acetyl-1,2-dideoxy-D-gly-

cero- β -L-glucio-heptofurano)[1,2-d]-2-imidazolidine (**3c**). — Conventional treatment of **2c** (0.3 g, 0.6 mmol) with pyridine (2 mL) and acetic anhydride (1.5 mL, 15.9 mmol) gave **3c** (0.304 g, 80%). Recrystallized from 90% ethanol, it had m.p. 130–132°, $[\alpha]_{\text{D}}^{18}$ -85.3° , $[\alpha]_{578}^{18}$ -89.2° , $[\alpha]_{546}^{18}$ -104.7° , $[\alpha]_{436}^{18}$ -210.9° , $[\alpha]_{365}^{18}$ -422.5° (*c* 0.65, pyridine); ν_{max} 2960, 2920, 2900 and 2890 (C–H), 1730 (C=O), 1470 and 1445 (C=C aromatic), 1250 and 1215 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table I.

Anal. Calc. for $\text{C}_{29}\text{H}_{31}\text{BrN}_2\text{O}_9\text{S}$: C, 52.48; H, 4.52; Br, 12.06; N, 4.22; S, 4.82. Found: C, 52.22; H, 4.54; Br, 11.69; N, 4.23; S, 5.23.

2-(Benzylthio)-4-(D-galacto-pentitol-1-yl)-1-phenylimidazole (**4a**). — (a). A solution of **1a** (ref. 4; 0.5 g, 1.5 mmol) in 90% ethanol (5 mL) was treated with benzyl chloride (0.2 mL, 1.5 mmol), boiled under reflux for 2.5 h, cooled, evaporated, an aqueous solution of the syrupy residue (0.136 g, 24%) extracted with ether (3 \times 25 mL) and the product crystallized from water. Recrystallized from 90% ethanol, it had m.p. 156–158°, $[\alpha]_{\text{D}}^{18}$ $+30.4^\circ$ (*c* 0.5, pyridine); $\lambda_{\text{max}}^{96\% \text{ EtOH}}$ 265 nm (ϵ_{mM} 6.00); ν_{max} 3380 (OH), 1590, 1485 and 1445 cm^{-1} (C=C aromatic).

Anal. Calc. for $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_5\text{S}$: C, 60.56; H, 5.80; N, 6.72; S, 7.69. Found: C, 60.34; H, 5.69; N, 6.72; S, 7.91. Periodate consumption: 4.15 mol.

(b) Compound **4a** was also obtained from a solution of **2a** (0.29 g, 0.7 mmol) in absolute ethanol (3.5 mL) that was treated with trifluoroacetic acid (0.4 mL). The mixture was boiled under reflux for 1 h, and the trifluoroacetic acid removed by repeated evaporation with ethanol under diminished pressure. An aqueous solution of the resulting syrup was washed with ethyl acetate (3 \times 22 mL), and the product (0.102 g, 35%) crystallized from water. Recrystallized from 90% ethanol, it gave pure **4a**.

2-(Benzylthio)-4-(penta-O-acetyl-D-galacto-pentitol-1-yl)-1-phenylimidazole (**5a**). — Conventional treatment of **4a** (0.1 g, 0.24 mmol) with pyridine (1.2 mL) and acetic anhydride (0.6 mL, 6.4 mmol) gave **5a** (0.144 g, 99%). Recrystallized from ethanol, it had m.p. 92–94°, $[\alpha]_{\text{D}}^{18}$ $+65.1^\circ$ (*c* 0.52, pyridine); $\lambda_{\text{max}}^{96\% \text{ EtOH}}$ 262 nm (ϵ_{mM} 5.50); ν_{max} 1740 (C=O), 1585 and 1490 (C=C aromatic), and 1230 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table III.

Anal. Calc. for $\text{C}_{31}\text{H}_{34}\text{N}_2\text{O}_{10}\text{S}$: C, 59.41; H, 5.46; N, 4.46; S, 5.11. Found: C, 59.71; H, 5.55; N, 4.44; S, 5.63.

2-(Benzylthio)-4-(D-galacto-pentitol-1-yl)-1-p-tolylimidazole (**4b**). — (a) A solution of **1b** (ref. 4; 1.02 g, 3 mmol) in 90% ethanol was treated with benzyl chloride (0.4 mL, 3 mmol), and the mixture was boiled under reflux for 3 h, cooled, and evaporated. A solution of the resulting syrup in water (20 mL) was washed with ethyl acetate (3 \times 20 mL), made neutral with NaHCO_3 , and kept in a refrigerator to crystallize (0.8 g, 63%). Recrystallization from 90% ethanol gave **4b**; m.p. 143–145, $[\alpha]_{\text{D}}^{19}$ $+18.6^\circ$, $[\alpha]_{578}^{19}$ 19.5° , $[\alpha]_{546}^{19}$ 22.2° , $[\alpha]_{436}^{19}$ 42.1° , $[\alpha]_{365}^{19}$ 77.2° (*c* 0.43, pyridine); $\lambda_{\text{max}}^{96\% \text{ EtOH}}$ 264 nm (ϵ_{mM} 6.00); ν_{max} 3360–3220 (OH), 3040 and 3010 (C–H aromatic), 2930, 2910 and 2880 (C–H), 1590, 1505 and 1490 cm^{-1} (C=C aromatic).

Anal. Calc. for $C_{22}H_{26}N_2O_5S$: C, 61.40; H, 6.05; N, 6.51; S, 7.44. Found: C, 61.08; H, 6.11; N, 6.47; S, 7.58. Periodate consumption: 4.15 mol.

(b) Compound **4b** was also prepared from a solution of **2b** (0.2 g, 0.46 mmol) in ethanol (3 mL) that was treated with trifluoroacetic acid (0.3 mL). The mixture was boiled under reflux for 1 h, and evaporated under diminished pressure. The resulting syrup was diluted with water (15 mL), and made neutral with 0.1M NaOH; **4b** (0.072 g, 36%) crystallized from this solution.

2-(Benzylthio)-4-(penta-O-acetyl-D-galacto-pentitol-1-yl)-1-p-tolylimidazole (5b). — Conventional treatment of **4b** (0.1 g, 0.23 mmol) with pyridine (0.75 mL) and acetic anhydride (0.7 mL, 7.42 mmol) gave **5b** (0.133 g, 89%), which was purified by dissolving in the minimal volume of ethanol, pouring into ice-water (75 mL), and keeping for 4 days at 0°. The resulting, amorphous solid was filtered off, and dried; m.p. 71–72°, $[\alpha]_D^{19} +69.4^\circ$, $[\alpha]_{578}^{19} 71.7^\circ$, $[\alpha]_{546}^{19} 82.6^\circ$, $[\alpha]_{436}^{19} 149.4^\circ$, $[\alpha]_{365}^{19} 250.3^\circ$ (*c* 0.54, pyridine); $\lambda_{max}^{96\% EtOH} 262$ nm ($\epsilon_{mM} 5.50$); ν_{max} 1750 (C=O), 1515, 1510 and 1495 (C=C aromatic), and 1220 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table III.

Anal. Calc. for $C_{32}H_{36}N_2O_{10}S$: C, 60.00; H, 5.62; N, 4.37; S, 5.00. Found: C, 59.82; H, 5.48; N, 4.30; S, 5.46.

2-(Benzylthio)-1-(p-bromophenyl)-4-(D-galacto-pentitol-1-yl)imidazole (4c). — (a) A solution of **1c** (ref. 4; 0.423 g, 1.27 mmol) in 90% ethanol (6 mL) was treated with benzyl chloride (0.13 mL, 1.04 mmol), and the mixture was boiled under reflux for 3 h, cooled, evaporated, and the resulting syrup dissolved in 96% ethanol. This solution was evaporated to a dry foam that crystallized from water (7 mL); yield 0.331 g, 64%. Recrystallization from 1:2 ethanol–water gave **4c**; m.p. 168–170°, $[\alpha]_D^{17} +16.2^\circ$, $[\alpha]_{578}^{17} 18.6^\circ$, $[\alpha]_{546}^{17} 19.5^\circ$, $[\alpha]_{436}^{17} 38.0^\circ$, $[\alpha]_{365}^{17} 70.2^\circ$ (*c* 0.50, pyridine); $\lambda_{max}^{96\% EtOH} 271$ nm ($\epsilon_{mM} 6.50$); ν_{max} 3350–3180 (OH), 2920 and 2850 (C–H), 1580 and 1505 cm^{-1} (C=C aromatic).

Anal. Calc. for $C_{21}H_{23}BrN_2O_5S$: C, 50.90; H, 4.64; Br, 16.16; N, 5.65; S, 6.46. Found: C, 50.56; H, 4.39; Br, 16.33; N, 5.79; S, 6.81. Periodate consumption: 4.07 mol.

(b) Compound **4c** was also prepared from a solution of **2c** (0.26 g, 0.52 mmol) in ethanol (4 mL) that was treated with trifluoroacetic acid to 10% concentration. The mixture was boiled under reflux for 1.5 h, cooled and evaporated under diminished pressure. The resulting syrup was dissolved in water (10 mL), and the solution made neutral with NaHCO_3 , and kept in a refrigerator to crystallize (0.204 g, 79%).

2-(Benzylthio)-1-(p-bromophenyl)-4-(penta-O-acetyl-D-galacto-pentitol-1-yl)imidazole (5c). — Conventional treatment of **4c** (0.129 g, 0.26 mmol) with pyridine (1.3 mL) and acetic anhydride (0.63 mL, 6.76 mmol) gave **5c** (0.144 g, 79%). Recrystallized from 96% ethanol, it had m.p. 103–105°, $[\alpha]_D^{19} +60.2^\circ$, $[\alpha]_{578}^{19} 60.4^\circ$, $[\alpha]_{546}^{19} 69.9^\circ$, $[\alpha]_{436}^{19} 126.6^\circ$, $[\alpha]_{365}^{19} 212.4^\circ$ (*c* 0.49, pyridine); ν_{max} 2980, 2940 and 2890 (C–H), 1730 (C=O), 1595 and 1460 (C=C aromatic), and 1215 cm^{-1} (C–O–C); $^1\text{H-n.m.r.}$ data are given in Table III.

Anal. Calc. for $C_{31}H_{33}BrN_2O_{10}S$: C, 52.77; H, 4.71; Br, 11.32; N, 3.97; S, 4.54. *Found*: C, 53.03; H, 4.74; Br, 11.40; N, 3.85; S, 4.88.

ACKNOWLEDGMENTS

We thank Dr. E. Román Galán for helpful discussions of the manuscript, and one of us (F.R.V.) thanks the Ministry of Education and Science of Spain for the award of a scholarship.

REFERENCES

- 1 F. GARCÍA GONZÁLEZ, J. FERNÁNDEZ-BOLAÑOS, AND F. J. LÓPEZ APARICIO, *Synthetic Methods for Carbohydrates*, *Am. Chem. Soc., Symp. Ser.*, 39 (1976) 207–226.
- 2 G. WEITZEL, F. SCHNEIDER, H. GUGLIELMI, F. SEIF, W. D. HIRSCHMANN, AND J. DURST, *Z. Physiol. Chem.*, 348 (1967) 1277–1284.
- 3 M. V. ÁLVAREZ, M. D. ASTUDILLO, E. RAMOS, F. SANZ, AND Z. L. ZUMEL, *An. Real Soc. Españ. Fis. Quím.*, 62B (1966) 729–736.
- 4 F. GARCÍA GONZÁLEZ, J. A. GALBIS PÉREZ, J. I. FERNÁNDEZ GARCÍA-HIERRO, AND J. FERNÁNDEZ-BOLAÑOS, *An. Quím.*, 75 (1979) 1002–1004.
- 5 D. HORTON AND J. D. WANDER, *J. Org. Chem.*, 39 (1974) 1859–1863.
- 6 K. IZUMI, *Carbohydr. Res.*, 77 (1979) 218–224.
- 7 B. COXON, *Carbohydr. Res.*, 8 (1968) 125–134.
- 8 B. COXON, *Methods Carbohydr. Chem.*, 6 (1972) 513–539.
- 9 F. V. BRUTCHER, JR. AND W. BAURER, JR, *J. Am. Chem. Soc.*, 84 (1962) 2233–2236.
- 10 N. K. RICHTMYER AND C. S. HUDSON, *J. Am. Chem. Soc.*, 64 (1942) 1612–1613.
- 11 M. BLANC-MUESSER, J. DEFAYE, AND D. HORTON, *Carbohydr. Res.*, 87 (1980) 71–86.
- 12 F. GARCÍA GONZÁLEZ, M. GÓMEZ GUILLÉN, J. A. GALBIS PÉREZ, P. ARECES BRAVO, AND E. ROMÁN GALÁN, *An. Quím.*, 76C (1980) 130–135.
- 13 J. M. BOBBITT, *Adv. Carbohydr. Chem.*, 11 (1956) 1–41.
- 14 P. F. FLEURY AND J. LANGE, *J. Pharm. Chim.*, 17 (1933) 107–113, 196–208.
- 15 J. FERNÁNDEZ-BOLAÑOS, J. A. GALBIS PÉREZ, AND F. ZAMORA, *An. Quím.*, in press.